

## 25. Camphorsulfonamide-Shielded, Asymmetric 1,4-Additions and Enolate Alkylations; Synthesis of a Southern Corn Rootworm Pheromone

Preliminary Communication<sup>1)</sup>

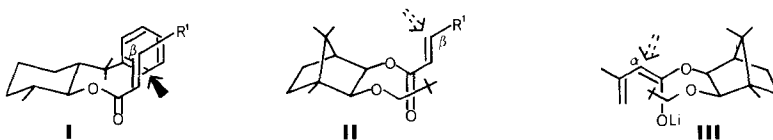
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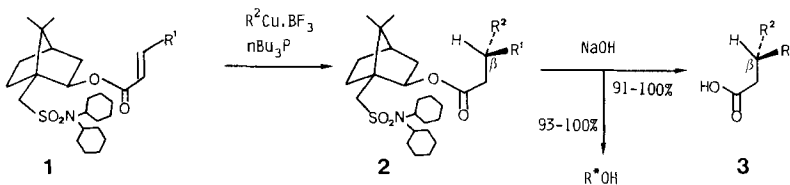
Using readily accessible 10-sulfonamido-isborneols as regenerable, chiral auxiliaries, highly face-selective C–C-bond formations at C<sub>α</sub> and C<sub>β</sub> of carboxylates could be conveniently achieved. Thus, conjugated additions of RCu to enoates (**1**→**2**) furnished, after saponification, β-substituted carboxylic acids **3** in 94–98% e.e. Similarly, propionates **12** yielded after deprotonation, enolate alkylation, and reductive ester cleavage the (*R*)-alcohols **15** in 78–98% e.e. The acid (+)-**3e** was converted to the pheromone (–)-**11**.

Recently, we have reported up to 99% π-face-selective, BF<sub>3</sub>·OEt<sub>2</sub>-mediated conjugate additions<sup>2)</sup> of organocopper reagents to chiral enoates **I** [2] and **II** [3].



The enolate-face shielding on C<sub>α</sub>-functionalization of the neopentyl ether **III** was comparatively less efficient (50% d.e.) [4]. Prompted by the practical utility of **1** (R<sup>1</sup>=H) as a dienophile in asymmetric *Diels-Alder* reactions<sup>3)</sup> [6], we studied the applicability of the camphorsulfonamide group as a chiral 1,4-acceptor- and enolate-auxiliary.

Scheme 1



<sup>1)</sup> Presented in part at the Autumn Meeting of the Swiss Chemical Society, Berne, October 19, 1984.

<sup>2)</sup> For alternative asymmetric additions of organocopper reagents see [1].

<sup>3)</sup> Review: [5].

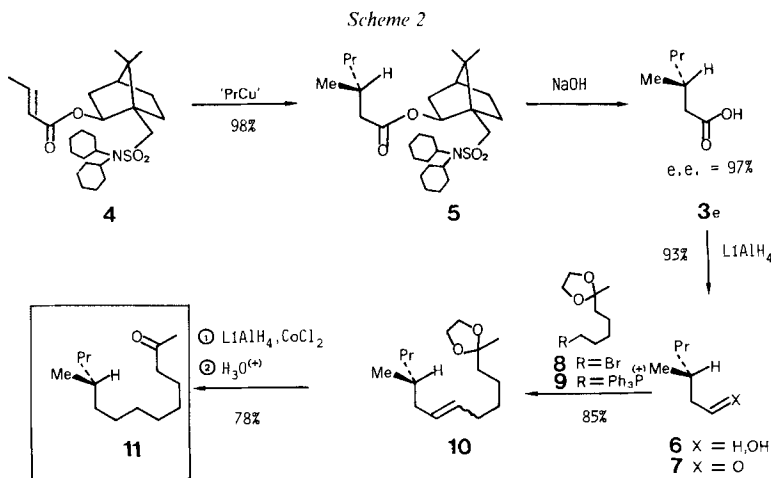
Table 1. Preparation of Chiral  $\beta$ -Substituted Carboxylic Acids **3** via Conjugate Addition **1**→**2**

Entry	R <sup>1</sup>	R <sup>2</sup>	Mol-equiv. R <sup>2</sup> -Cu <sup>a</sup> )	Yield [%] of <b>2</b>	e.e. % of <b>3</b>
a	Me	Pr	2	98	95 (97)
b	Me	Bu	2	89	97
c	Me	Vinyl	10	80	98
d	Me	2-Propenyl	10	84	94
e	Pr	Me	10	89	94
f	Bu	Me	10	93	97

<sup>a</sup>) i) Addition of R<sup>2</sup>Li (4 mmol; see *Footnote 6*) to CuI · P(Bu)<sub>3</sub> complex in Et<sub>2</sub>O at -78°, stirring for 30 min → -30°. ii) Addition of BF<sub>3</sub> · OEt<sub>2</sub> (4 mmol) at -78°; iii) Addition of enoate **1** (0.4–2 mmol) in Et<sub>2</sub>O/THF 4:1 at -78°, warming up to -40° over 5 to 20 h. iv) Addition of sat. aq. NH<sub>4</sub>Cl/Et<sub>2</sub>O, stirring of the Et<sub>2</sub>O phase with MCPBA (4 mmol) for 10 min and workup.

As indicated in *Scheme 1* and *Table 1*, treatment of crotonates **1a–d**<sup>4)</sup> with RCu · PBu<sub>3</sub> · BF<sub>3</sub><sup>6)</sup> at -78°C → -40°C in Et<sub>2</sub>O/THF ~ 15:1 furnished the 1,4-adducts **2a–d** in good-to-excellent yields. Saponification of **2** (1N NaOH, in aq. EtOH, reflux) gave the  $\beta$ -substituted carboxylic acids **3a–d** in 94–98% e.e.<sup>7)</sup> with virtually complete recovery of the crystalline auxiliary. The sense of induction at C <sub>$\beta$</sub>  of **3** was readily reversed either by interchanging R<sup>1</sup> and R<sup>2</sup> (see *Table 1*, examples a/e, b/f) or by using the antipodal inductor group [6].

The acid **3e**<sup>4)</sup>, obtained in 97% e.e. via the sequence **4**<sup>4)</sup>→**5**<sup>4)</sup>→**3e** (or in 94% e.e. from **1e**) served as a key intermediate for the synthesis of the southern corn rootworm phe-



<sup>4</sup>) All new compounds were characterized by IR, <sup>1</sup>H-NMR and MS.

<sup>5</sup>) The chiral esters **1**, **4** and **12** were prepared [7] by heating a mixture of the chiral alcohol (1 mol-equiv.), AgCN (1.4 mol-equiv.) and the corresponding acid chloride (2.0 mol-equiv.) in benzene at reflux for 4 h under Ar.

<sup>6</sup>) The starting reagents MeLi, PrLi, CH<sub>2</sub>=CHLi and CH<sub>2</sub>=CH(CH<sub>3</sub>)Li were prepared by metalation of MeI, PrBr, vinyl chloride and 2-propenyl bromide with a lithium dispersion (3 mol-equiv.) in Et<sub>2</sub>O (THF for vinyl chloride) using a *Vibromix*.

<sup>7</sup>) The enantiomeric purities of carboxylic acids **3** were determined by HPLC analyses of their (*S*)- $\alpha$ -naphthyl-ethylamides [8] and their absolute configurations established by chiroptic comparison.

romone **11** [9]<sup>8)</sup> (Scheme 2). Reduction of **3e** with LiAlH<sub>4</sub> (2 mol-equiv. Et<sub>2</sub>O, 0° → +20°, 2 h) followed by oxidation [11] of the alcohol **6<sup>4</sup>** ((COCl)<sub>2</sub>, DMSO, -50 to -60°, 1 h) gave aldehyde **7<sup>4</sup>** (80%) which on Wittig reaction (add BuLi (1 mmol) to phosphonium bromide **9<sup>9</sup>** (1.2 mmol) in THF at -78° → +20° → -78°, add **7** (0.5 mmol) at -78° → +20°) yielded the olefin **10<sup>4</sup>**. Hydrogenation of **10** under non-epimerizing conditions<sup>10)</sup> (add LiAlH<sub>4</sub> (0.7 mmol) to a solution of dry CoCl<sub>2</sub> (0.7 mmol) and ketal **10** (0.3 mmol) in THF at -78°, stir at +20°, 24 h) [14] and subsequent acetal cleavage (HOAc/H<sub>2</sub>O, 4:1, 50°, 30 min) afforded the pheromone **11<sup>4</sup>** in high enantiomeric purity ([α]<sub>D</sub><sup>24</sup> = -1.61° (c = 4.1, CHCl<sub>3</sub>); [10a]: [α]<sub>D</sub><sup>24</sup> = -1.71° (c = 8.6, CHCl<sub>3</sub>)) identified by comparison (IR, <sup>1</sup>H-NMR and MS) with the published spectral data of **11** [10].

The versatility of camphorsulfonamides as practical π-face-shielding elements is further exemplified by the asymmetric enolate alkylations<sup>11)</sup> presented in Scheme 3 and Table 2.

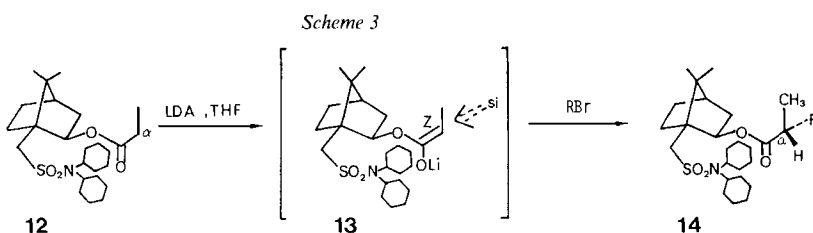
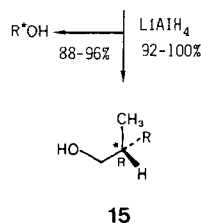


Table 2. Asymmetric Alkylation/Reduction **12** → **14** → **15**

Entry	R	Purification of <b>14</b> (m.p. °C)	Yield% of <b>14</b>	e.e. % of <b>15</b>
a	PhCH <sub>2</sub>	crude	84	89
		1 cryst. (179–81)	61	98
b	CH <sub>2</sub> =CH-CH <sub>2</sub>	crude	94	88
c	Pr <sup>2)</sup>	crude	92	78

a) Add a solution of PrBr in THF/HMPA (3 mol-equiv.) to enolate **13**.



Kinetically controlled deprotonation [16] of the propionate **12<sup>4</sup>** (LDA (1.1. mol-equiv.), THF, -78°) followed by addition of a primary bromide to the enolate **13** gave the chiral α-substituted esters **14<sup>4</sup>** in 84–94% yields and with 78 to 89% diastereoface differentiation. The diastereomeric purity of **14a** was raised to 98% d.e. by simple crystallization. Notably, even the non-activated PrBr led to the alkylation product **14c** in 92% yield (78% d.e.). Reductive cleavage (LiAlH<sub>4</sub> (2 mol-equiv.), Et<sub>2</sub>O, 0–20°, 30 min) of

<sup>8)</sup> For other syntheses of the pheromone **11** see [10].

<sup>9)</sup> Bromide **8<sup>4</sup>** was prepared by treatment of 1-bromo-6-heptene with Hg(OAc)<sub>2</sub> + TsOH in ethylene glycol/THF (30 min, +20°), followed by addition of PdCl<sub>2</sub>, LiCl, LiCO<sub>3</sub>, CuCl<sub>2</sub> (heating at reflux for 2 h) [12]. A mixture of bromide **8** and PPh<sub>3</sub> was slowly warmed up to 160° and kept at this temperature for 6 h to give the salt **9** (90%).

<sup>10)</sup> Under these conditions, (*R*)-citronellal was hydrogenated without epimerization, whereas considerable racemization occurred on hydrogenation with Pd/C, EtOH, H<sub>2</sub> (1 atm, 20°) [13]. An analogous epimerization may account for the relatively low optical rotation reported for synthetic **11** [10b].

<sup>11)</sup> For other asymmetric enolate alkylations see [15].

the alkylated esters **14** gave the unchanged auxiliary and the (*R*)-alcohols **15** in 78–98% e.e.<sup>12)</sup>.

The asymmetric  $\alpha$ - and  $\beta$ -functionalizations of esters, described above, are currently the subject of further investigation in this laboratory.

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## REFERENCES

- [1] G. H. Posner, in 'Asymmetric Synthesis', Ed. J. D. Morrison, Academic Press, New York, 1983, Vol. 2, p. 225; F. Leyendecker, F. Jesser, D. Laucher, *Tetrahedron Lett.* **1983**, 24, 3513.
- [2] W. Oppolzer, H. Löher, *Helv. Chim. Acta* **1981**, 64, 2808.
- [3] W. Oppolzer, R. Moretti, T. Godel, A. Meunier, H. Löher, *Tetrahedron Lett.* **1983**, 24, 4971.
- [4] W. Oppolzer, R. Pitteloud, G. Bernardinelli, K. Baettig, *Tetrahedron Lett.* **1983**, 24, 4975; W. Oppolzer, in 'Selectivity – a Goal for Synthetic Efficiency', Ed. W. Bartmann and B. M. Trost, Verlag Chemie, Weinheim, 1984, p. 137.
- [5] W. Oppolzer, *Angew. Chem.* **1984**, 96, 840; *ibid. Int. Ed.* **1984**, 23, 876.
- [6] W. Oppolzer, C. Chapuis, G. Bernardinelli, *Tetrahedron Lett.* **1984**, 25, 5885.
- [7] S. Takimoto, J. Inanaga, T. Katsuki, M. Yamaguchi, *Bull. Chem. Soc. Jpn.* **1976**, 49, 2335.
- [8] W. H. Pirkle, J. R. Hauske, *J. Org. Chem.* **1977**, 42, 1839.
- [9] P. L. Guss, J. H. Tumlinson, P. E. Sonnet, J. R. Laughlin, *J. Chem. Ecol.* **1983**, 9, 1363.
- [10] a) P. E. Sonnet, *J. Org. Chem.* **1982**, 47, 3793; b) S. Senda, K. Mori, *Agric. Biol. Chem.* **1983**, 47, 795.
- [11] A. J. Mancuso, S. L. Huang, D. Swern, *J. Org. Chem.* **1978**, 43, 2480.
- [12] D. F. Hunt, G. T. Rodehoever, *Tetrahedron Lett.* **1972**, 3595.
- [13] W. Oppolzer, R. Moretti, Autumn Meeting of the Swiss Chemical Society, Berne, October 19, 1984.
- [14] E. C. Ashby, J. J. Lin, *J. Org. Chem.* **1978**, 43, 2567.
- [15] D. A. Evans, in 'Asymmetric Synthesis', Ed. J. D. Morrison, Academic Press, New York, 1984, Vol. 3, p. 1.
- [16] R. E. Ireland, R. H. Muller, A. K. Willard, *J. Am. Chem. Soc.* **1976**, 98, 2868.

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<sup>12)</sup> As expected, deprotonation of **12** with LDA in THF/HMPA 4:1 followed by addition of benzyl bromide and subsequent crystallization and reduction furnished the enantiomer of alcohol **15a** in 80% e.e. The enantiomeric purity of the alcohols **15** was determined by HPLC analysis [8] either of their carbamates derived from (*R*)- $\alpha$ -naphthylethylisocyanate (entry a, *Table 2*) or of the (*S*)- $\alpha$ -naphthylethylamides obtained from **15** by successive *Jones'* oxidation and amidation (entries b, c).